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# Phosphorus, Sulfur, and Silicon and the Related Elements

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# 1,1-Bis(Methylthio)-2-Nitroethene, a Versatile Synthon for the Synthesis of Oximinoorthothioesters

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1,1-BIS(METHYLTHIO)-2-NITROETHENE, A VERSATILE SYNTHON FOR THE SYNTHESIS OF OXIMINOORTHOTHIOESTERS.

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Abstract 1,1-bis(methylthio)-2-nitroethene gives a stable hydro-xynitrilium ion in trifluoromethanesulfonic acid. This cation can be trapped by various nucleophiles to yield  $\alpha$ -oximinoorthothio-esters.

#### INTRODUCTION

Nitroderivatives and nitronates are polyprotonated in superacidic media to give transient species which can be trapped with suitable nucleophiles such as aromatics to gives oximes <sup>1,2</sup>. The stereochemistry of the resulting product -in which the aryl and hydroxyl group are cis- is consistent with the intermediacy of an hydroxynitrilium ion <sup>3,4</sup>.

$$R-CH_2-NO_2$$
  $\longrightarrow$   $R-C=N-OH$   $\longrightarrow$   $R-C=N$ 

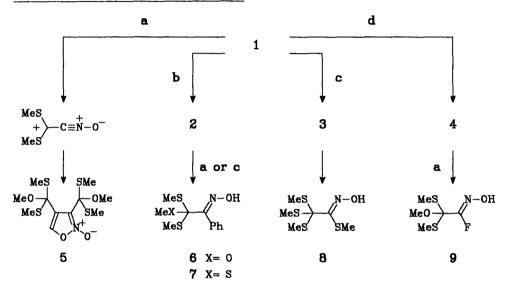
#### HYDROXYNITRILIUM ION

Protonation of the title compound gives transient cations and finally the sole hydroxynitrilium ion 1 which has been fully characterized by  $^{1}{\rm H}$  and  $^{13}{\rm C}$  NMR spectroscopy in TFSA at low temperature  $^{5}$ .  $^{13}{\rm C}$  chemical shifts are in good agreement with what has been reported for nitrile N-oxides  $^{6}$ .

The stabilizing effect of sulfur atom accounts for the specific reactivity of 1 compared to protonated forms of substituted nitroethylenes in TFSA $^7$ . Ion 1 is easily trapped on the hydroxynitrilium group to give ions2 to 4. The resulting 1,3-dithioallylic cationic system is very stable in TSFA or HF-SbF $_5$  and can only be trapped by nucleophiles when

the acidity is destroyed by quenching. The yield in products is good to excellent (65 to < 90%).

## REACTIVITY OF HYDROXYNITRILIUM ION



a = MeOH in excess; b = PhH in excess; c = MeSH in excess; d = HF-SbF $_5$  Isomerization of orthothioesters in TFSA allows group exchange reaction for example 7 ionizes to 2 and trapping with methanol yields 7 (65%). All the ions described here have been fully characterized by  $^1$ H and  $^{13}$ C NMR spectroscopy at low temperature  $^8$ .

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